

Triaqua(18-crown-6)calcium(II) tetrachloro-  
zincate(II)Xi Liu<sup>a\*</sup> and Guo-Cong Guo<sup>b</sup><sup>a</sup>College of Chemistry, Chongqing Normal University, Chongqing 400047, People's Republic of China, and <sup>b</sup>State Key Laboratory of Structural Chemistry, Fujian Institute of Research on the Structure of Matter, Chinese Academy of Sciences, Fuzhou, Fujian 350002, People's Republic of China

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## Key indicators

Single-crystal X-ray study  
 $T = 293$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.005$  Å  
 $R$  factor = 0.040  
 $wR$  factor = 0.108  
Data-to-parameter ratio = 22.4For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.Diffusion of diethyl ether into a reaction mixture of zinc salts, crown ether and calcium chloride in tetrahydrofuran (THF) led to the title compound,  $[\text{Ca}(\text{C}_{12}\text{H}_{24}\text{O}_6)(\text{H}_2\text{O})_3][\text{ZnCl}_4]$ . The cation is linked to the anion *via*  $\text{O}-\text{H}\cdots\text{Cl}$  hydrogen bonding.Received 9 December 2006  
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## Comment

The ability of 18-crown-6 ether (18-C-6) to form complexes with different metal ions has been widely investigated. We report here the synthesis and crystal structure of the title compound, (I).

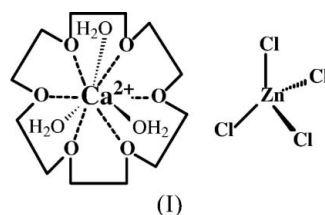
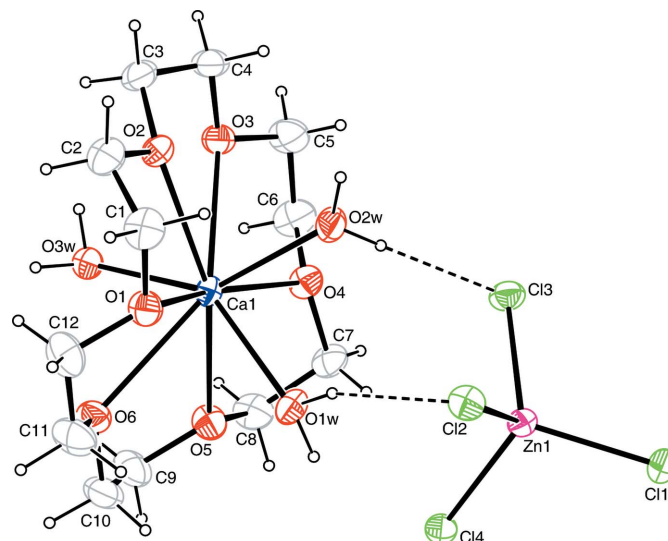
The crystal structure of (I) consists of  $[\text{Ca}(18\text{-C-6})(\text{H}_2\text{O})_3]^{2+}$  cations and  $[\text{ZnCl}_4]^{2-}$  anions. The structure of the cation is similar to that found in  $[\text{Ca}(18\text{-C-6})(\text{H}_2\text{O})_3][\text{Cu}_5\text{I}_7]$  (Nurtaeva & Holt, 2002). The crystal structure of the  $[\text{ZnCl}_4]^{2-}$  anion is in good agreement with that in reported complexes (Jackson *et al.*, 1981; Smolenaers *et al.*, 1981; Otter & Hartshorn, 2004). The cation is linked to the anion *via*  $\text{O}-\text{H}\cdots\text{Cl}$  hydrogen bonding (Table 1).

Figure 1

The molecular structure of (I) with 30% probability displacement ellipsoids (arbitrary spheres for H atoms). Dashed lines indicate hydrogen bonds.

Experimental

ZnCl<sub>2</sub>·2H<sub>2</sub>O (35 mg, 0.2 mmol), 18-C-6 (53 mg, 0.2 mmol) and CaCl<sub>2</sub> (22 mg, 0.2 mmol) were added to THF (10 ml) and the reaction mixture was stirred at 333 K for 6 h. The mixture was filtered and the filtrate was reduced to 5 ml in a small tube, which was loaded into a large vial containing 5 ml diethyl ether. The large vial was sealed and left undisturbed at room temperature; colorless crystals of (I) were obtained after 7 d (yield: 76%). Calculated for C<sub>12</sub>H<sub>30</sub>CaCl<sub>4</sub>O<sub>9</sub>Zn: C 25.48, H 5.35%; found: C 25.60, H 5.42%.

Crystal data

[Ca(C<sub>12</sub>H<sub>24</sub>O<sub>6</sub>)(H<sub>2</sub>O)<sub>3</sub>][ZnCl<sub>4</sub>] Z = 4  
*M<sub>r</sub>* = 565.61 *D<sub>x</sub>* = 1.563 Mg m<sup>-3</sup>  
 Monoclinic, *P*2<sub>1</sub>/*c* Mo *K*α radiation  
*a* = 12.573 (3) Å μ = 1.72 mm<sup>-1</sup>  
*b* = 9.489 (2) Å *T* = 293 (2) K  
*c* = 20.546 (5) Å Prism, colorless  
 β = 101.262 (3)° 0.30 × 0.25 × 0.25 mm  
*V* = 2404.0 (11) Å<sup>3</sup>

Data collection

Rigaku Mercury CCD 17093 measured reflections  
 diffractometer 5468 independent reflections  
 ω scans 4955 reflections with *I* > 2σ(*I*)  
 Absorption correction: multi-scan *R*<sub>int</sub> = 0.035  
 (*CrystalClear*; Rigaku, 2002) θ<sub>max</sub> = 27.5°  
*T*<sub>min</sub> = 0.515, *T*<sub>max</sub> = 0.655

Refinement

Refinement on *F*<sup>2</sup> H-atom parameters constrained  
*R*[*F*<sup>2</sup> > 2σ(*F*<sup>2</sup>)] = 0.040 *w* = 1/[σ<sup>2</sup>(*F*<sub>o</sub><sup>2</sup>) + (0.06*P*)<sup>2</sup> + *P*]  
*wR*(*F*<sup>2</sup>) = 0.108 where *P* = (*F*<sub>o</sub><sup>2</sup> + 2*F*<sub>c</sub><sup>2</sup>)/3  
*S* = 1.00 (Δ/σ)<sub>max</sub> = 0.002  
 5468 reflections Δρ<sub>max</sub> = 0.34 e Å<sup>-3</sup>  
 244 parameters Δρ<sub>min</sub> = -0.39 e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
O1W–H1E···Cl4	0.93	2.34	3.227 (2)	158
O1W–H1F···Cl2	0.86	2.48	3.306 (2)	161
O2W–H2E···Cl3	0.92	2.35	3.236 (2)	163
O2W–H2F···Cl2 <sup>i</sup>	0.99	2.33	3.285 (2)	163
O3W–H3E···Cl4 <sup>ii</sup>	0.85	2.67	3.410 (2)	146
O3W–H3F···Cl1 <sup>ii</sup>	0.96	2.22	3.157 (2)	165

Symmetry codes: (i) -*x* + 1, -*y* + 1, -*z* + 1; (ii) *x*, -*y* + 3/2, *z* - 1/2.

Water H atoms were located in a difference Fourier map and refined as riding in their as-found relative positions, with *U*<sub>iso</sub>(H) = 1.5*U*<sub>eq</sub>(O). Other H atoms were placed in calculated positions, with C–H = 0.97 Å, and refined in riding mode, with *U*<sub>iso</sub>(H) = 1.2*U*<sub>eq</sub>(C).

Data collection: *CrystalClear* (Rigaku, 2002); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXTL* (Siemens, 1994); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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